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UNIT 3: Ionic Bonding Atoms can form ions because all atoms want full s and p subshells. A full s and p subshell (of the same energy level) is also called a "complete octet", and is a very stable electron arrangement. Atoms therefore lose or gain their valence electrons to achieve the electron configuration of the nearest noble gas.

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The notes in this article can be used to study smaller portions of the curriculum or to review for the final AP Chemistry exam. As of the 2019 updates, there are nine units that organize all the concepts in the course, so I've categorized these notes according to that framework.

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Energy can "flow" or be "transferred" from one system to another and so cause changes. 3. Energy maintains its identity after being transferred.

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